

Hydroformylation with Integrated SILP Catalyst-Membrane Separation Reaction System

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Hydroformylation as large volume process

Hydroformylation is one of the largest homogeneously catalyzed processes in the chemical industry. Catalysis with supported ionic liquid phase (SILP) catalyst systems (Fig. 1) has successfully been established over the last decades as an industrially attractive approach conducting liquid- and gas-phase reactions.¹ Hydroformylation of olefins by syngas (Fig. 2) to produce aldehydes is a highly important and frequently studied catalytic reaction. Several SILP catalyst systems have demonstrated industrial potential.²

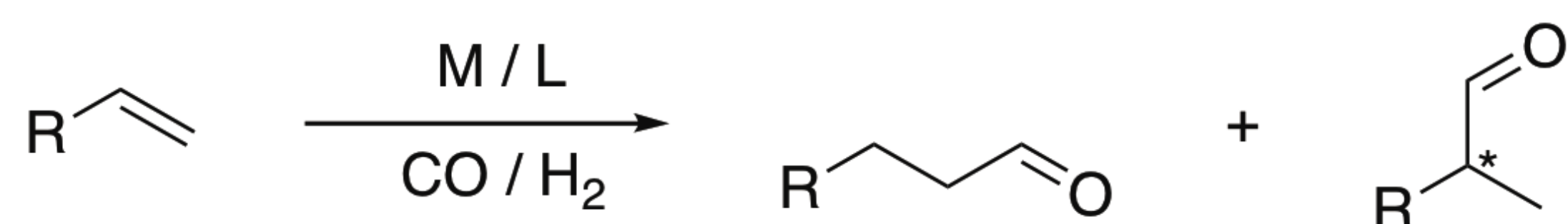
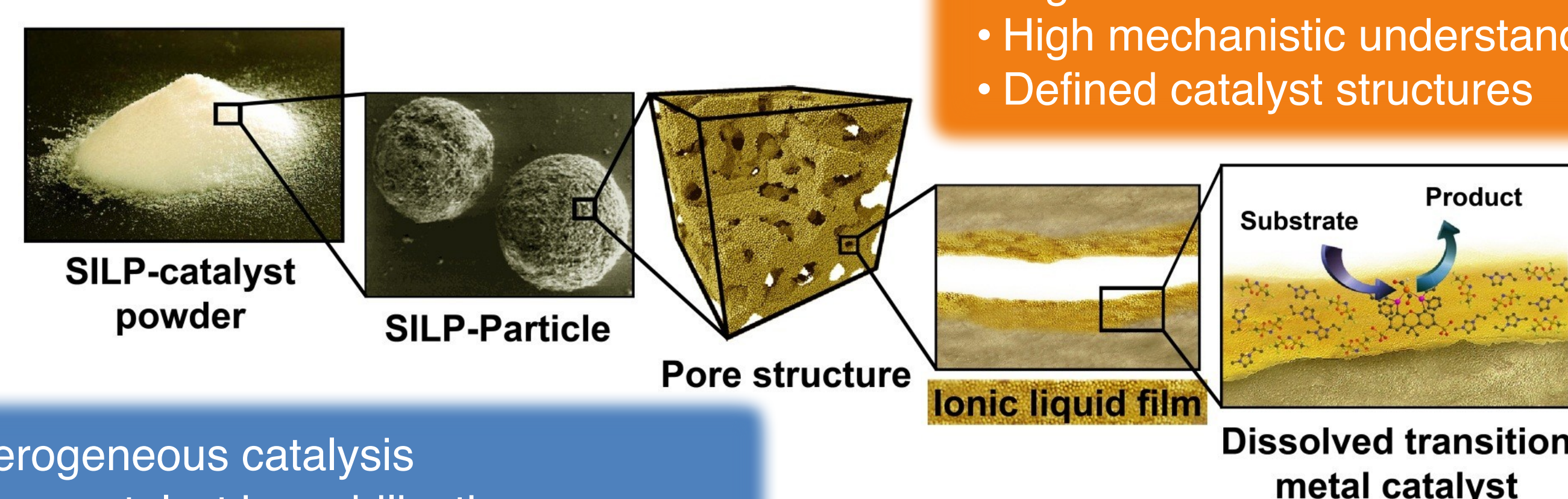


Fig. 2: Metal catalyzed hydroformylation of olefins by syngas



Homogeneous catalysis

- High activities and selectivities
- High mechanistic understanding
- Defined catalyst structures

Heterogeneous catalysis

- Easy catalyst immobilization
- Easy product recovery
- Usually simplified technical processes

Fig. 1: Supported Ionic Liquid Phase (SILP) materials build the bridge between homogeneous and heterogeneous catalysis

Drawback in SILP hydroformylation

Long-term catalytic performance test (Fig. 3) of SILP catalyst on hydroformylation of olefins exhibits decreasing activity due to the formation of “heavies” by undesired condensation reactions (Fig. 4). This hampers both, activity and selectivity and prevents the implementation of SILP in industry.

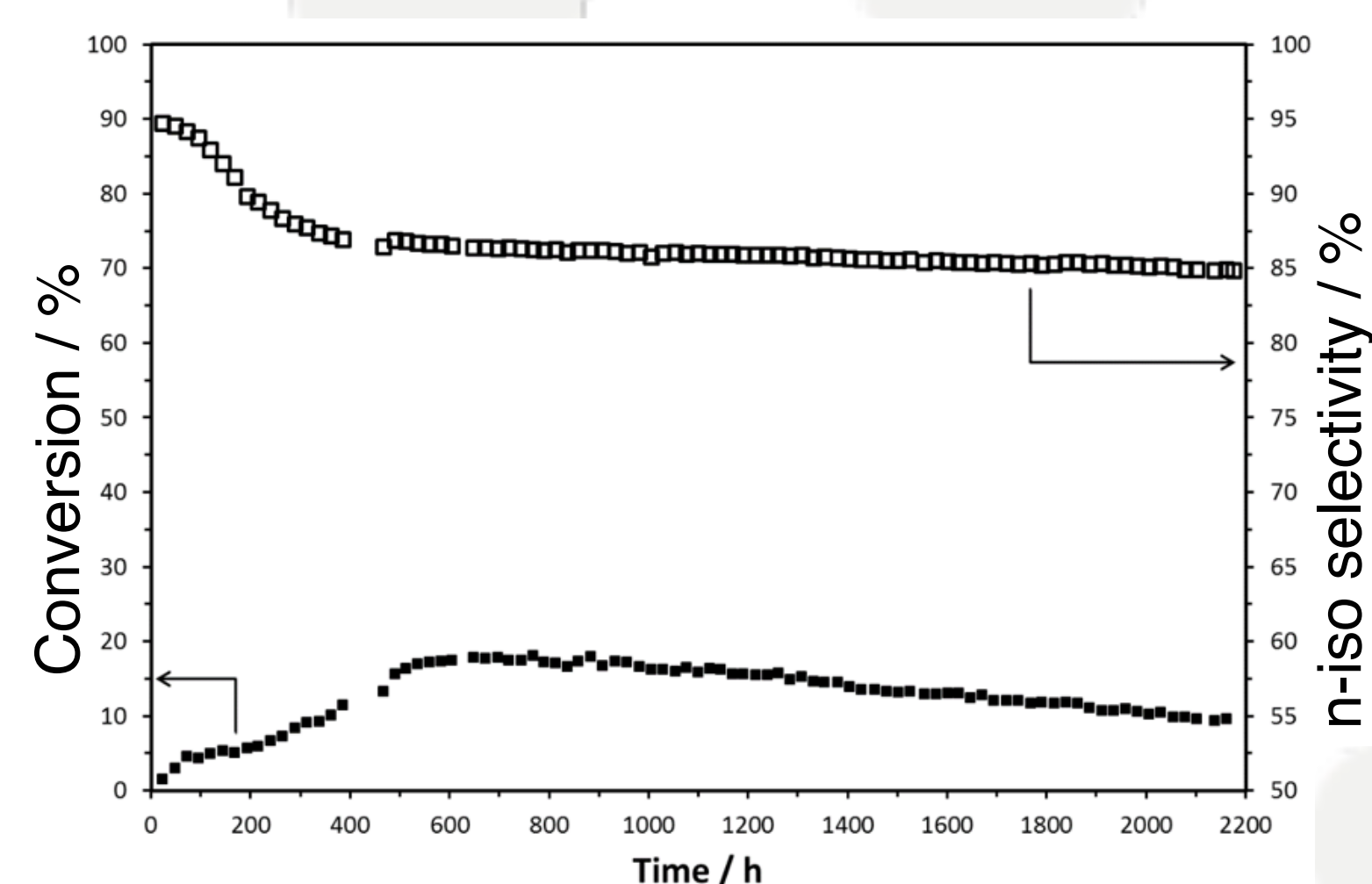


Fig. 3: Long-term performance test of SILP materials

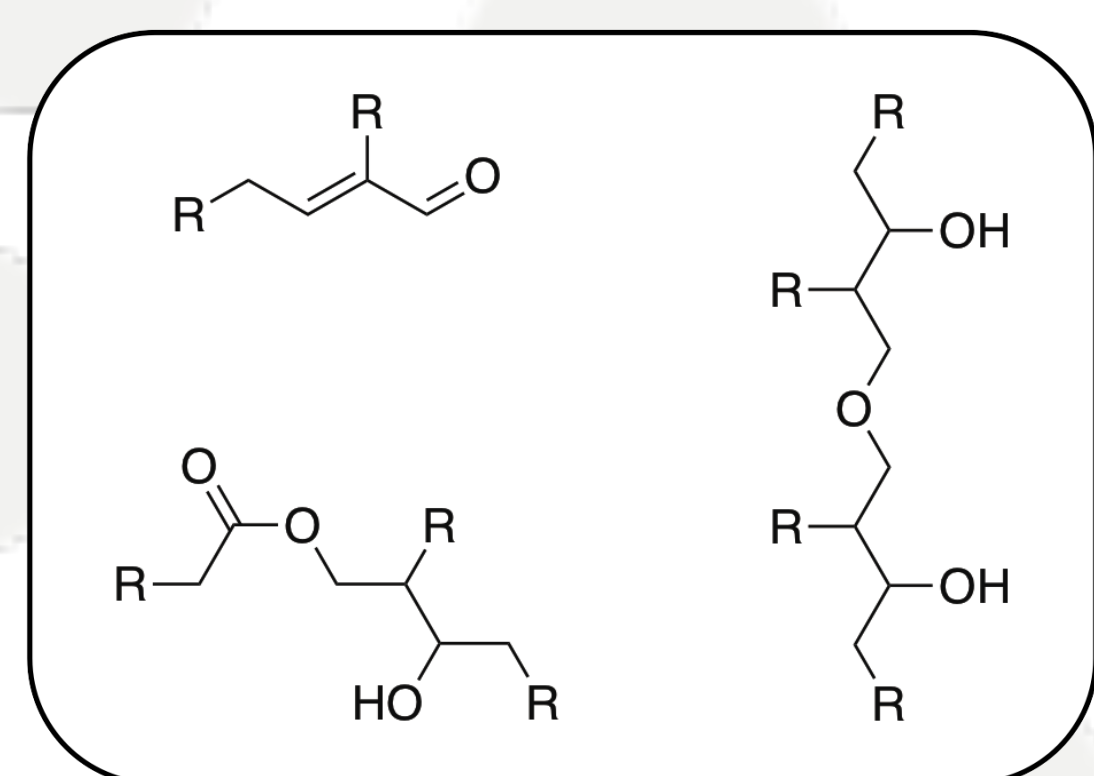


Fig. 4: Possible formation of “heavies” during hydroformylation

ROMEEO: Reactor Optimization by Membrane Enhanced Operation

ROMEEO's two-in-one reactor design (Fig. 5) improves long-term stability by avoiding aldol condensation. This is achieved by a membrane coating enabling continuous and selective (by-) product separation from the reaction.

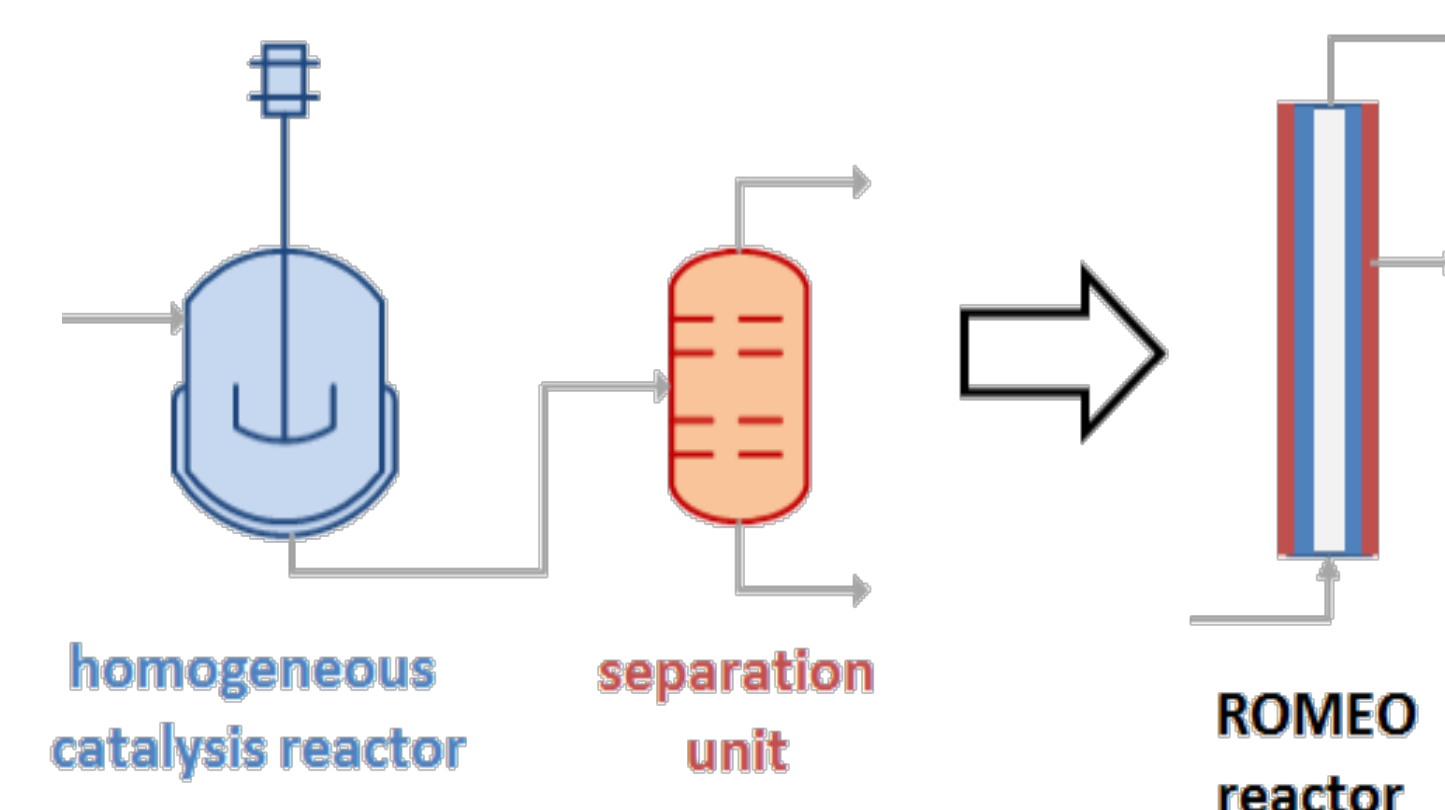


Fig. 5: ROMEEO's reactor combines two essential process steps

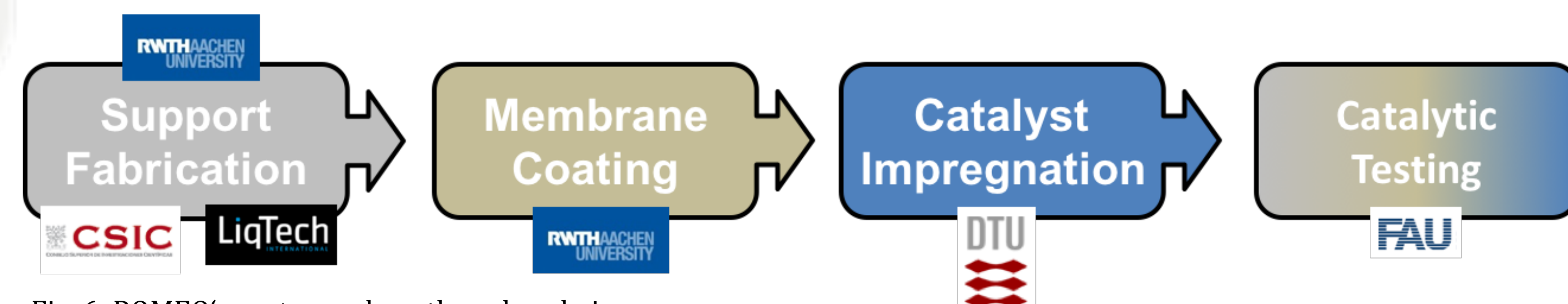


Fig. 6: ROMEEO's partners along the value chain

Results

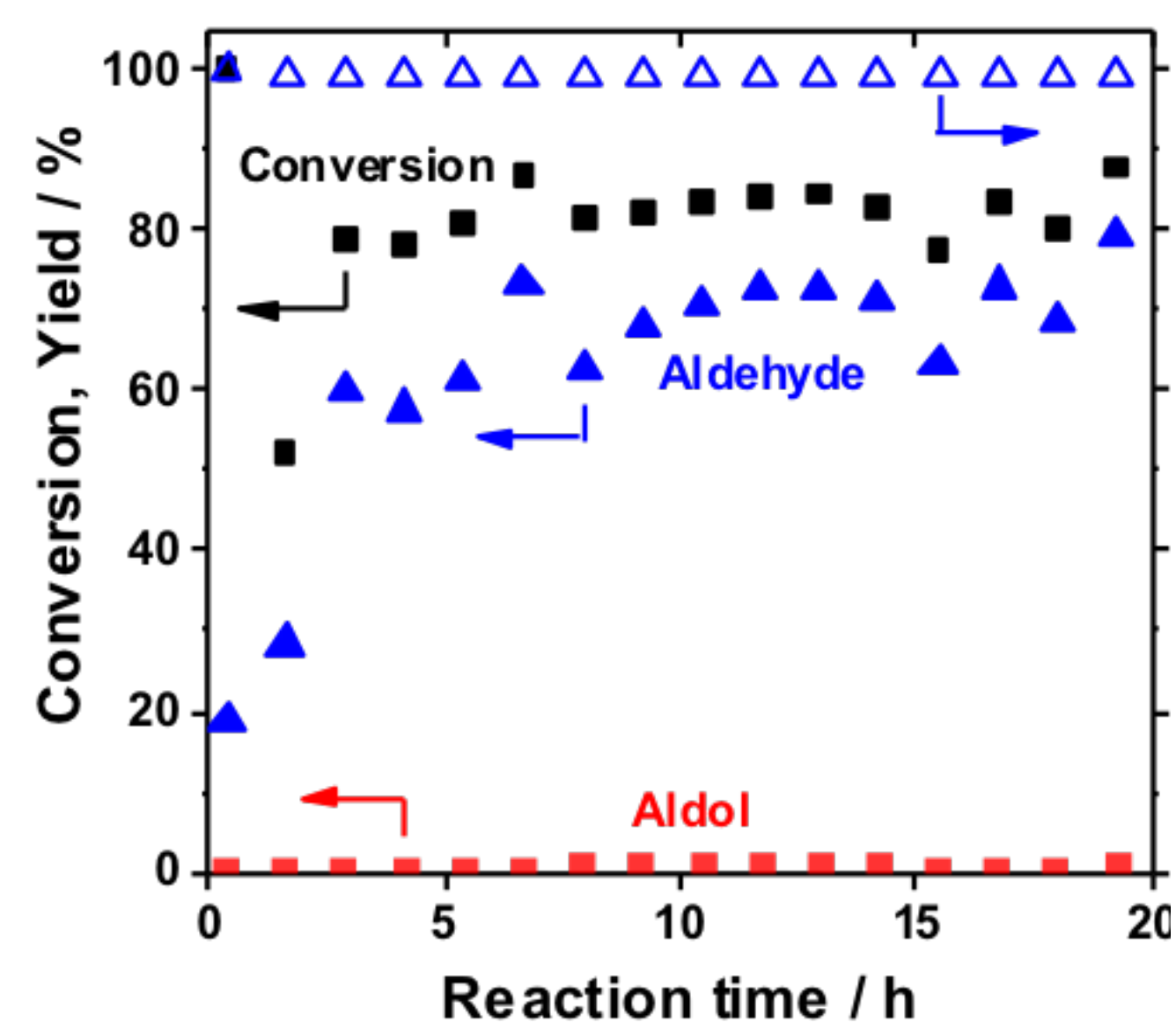


Fig. 7: Gas-phase hydroformylation with ROMEEO reactor (without membrane)

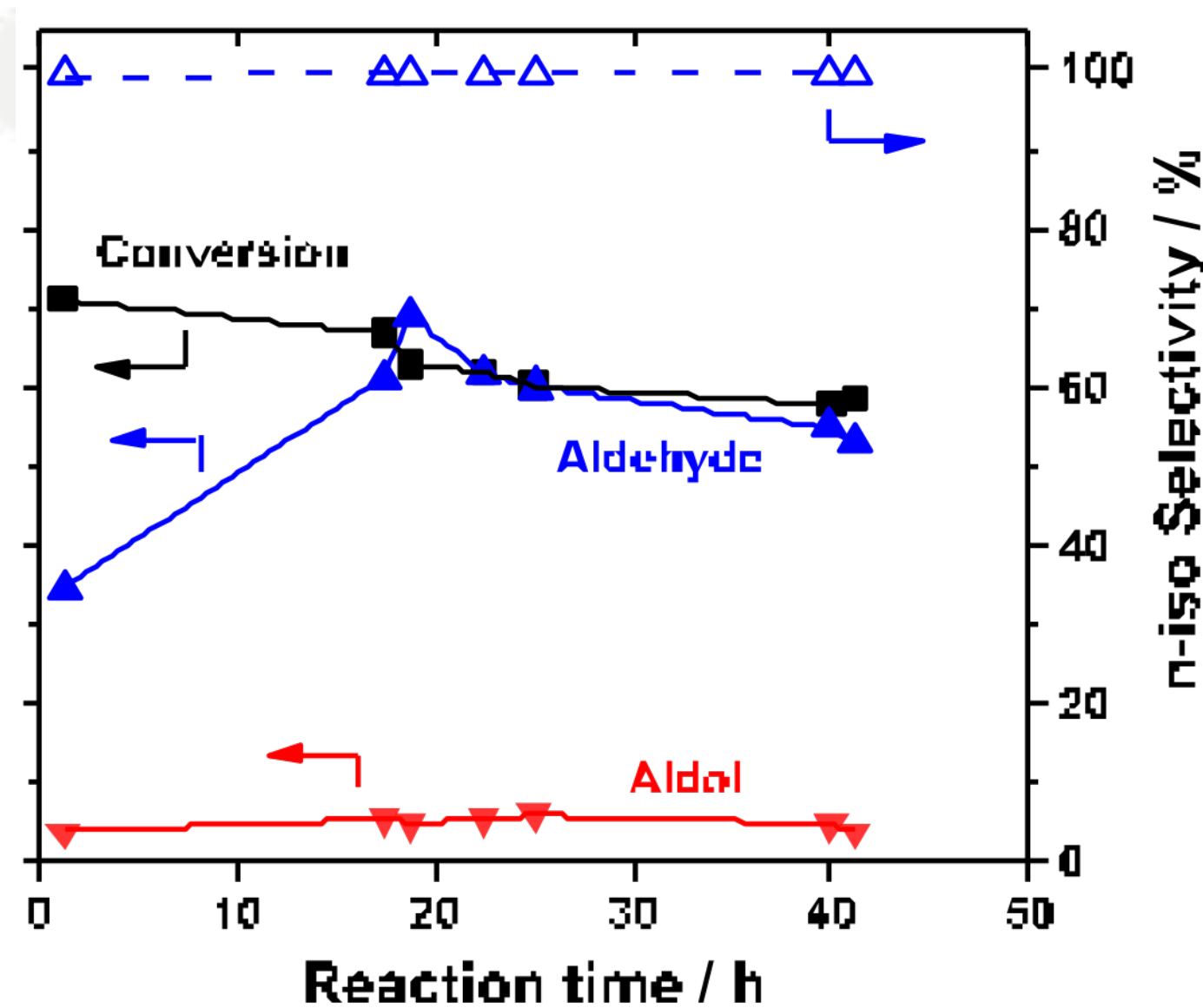


Fig. 8: Gas-phase hydroformylation with ROMEEO reactor (with membrane - harmed during operation)

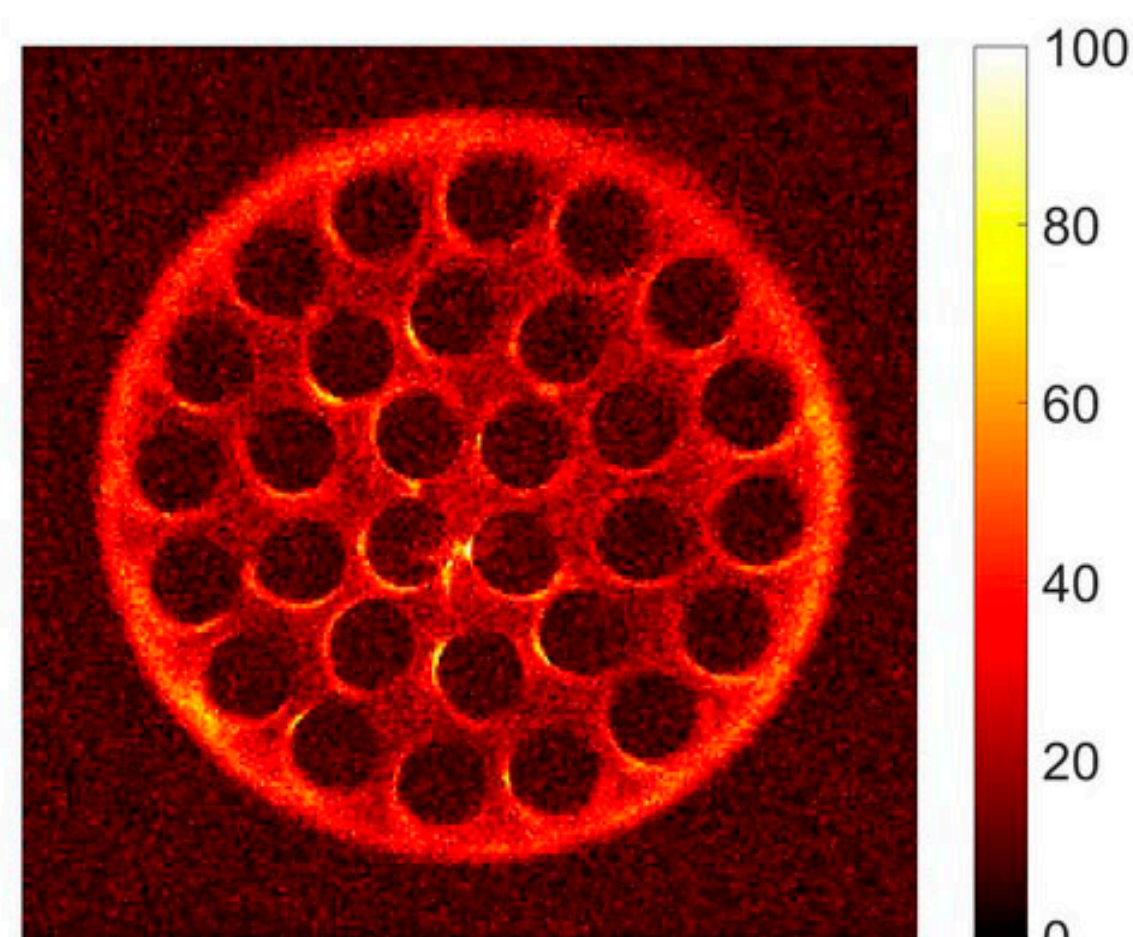


Fig. 9: Magnetic resonance image (MRI) of SILP impregnated ceramic support

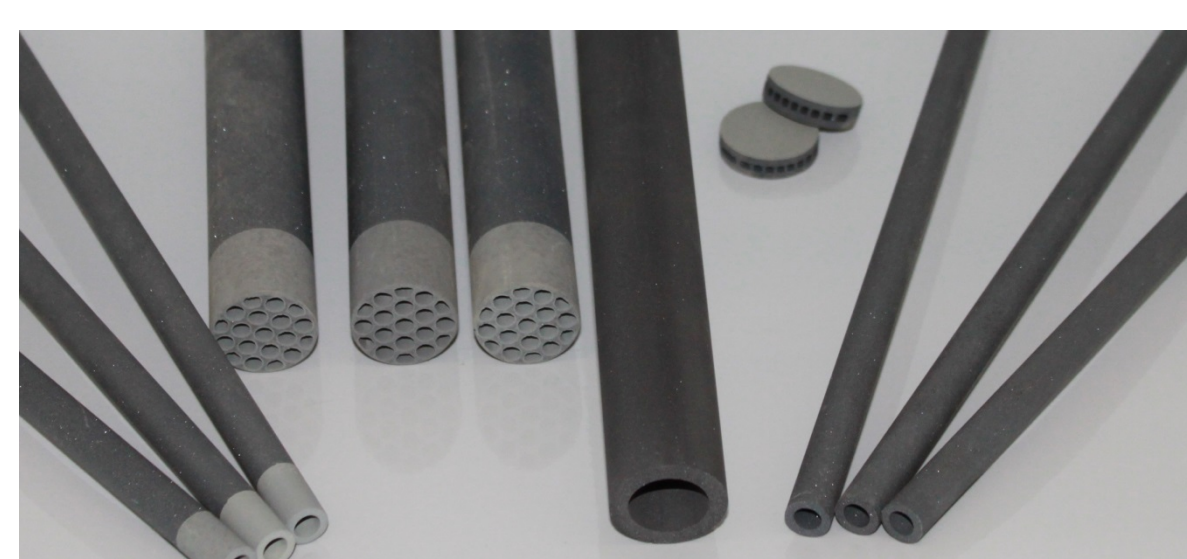


Fig. 10: Various ceramic supports

Conclusions

In summary, preliminary results indicate that ROMEEO's “two-in-one” reactor concept has a great potential to be applied in the chemical industry. At this point, the mechanical stability of the membrane needs to be improved. If this is achieved, a reduction in emissions, energy consumption, space needs and costs in the chemical industry are expected with this new reactor type.



Fig. 11: A first-class consortium of nine partners from five European countries make up ROMEEO's team

References

- [1] S. Bhaduri, D. Mukesh. Homogeneous Catalysis: Mechanisms and Industrial Applications, 2nd Edition. John Wiley & Sons, 2014.
- [2] R. Franke, D. Selent, Armin Börner. Applied Hydroformylation. Chemical Reviews, 2012, 112 (11), pp 5675–5732.

Acknowledgements



This project has received funding from the European Union's Horizon 2020 research and innovation programme under grant agreement No 680395